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 இலங்கைப் பரீட்சைத் திணைக்களம் இலங்கைப் பரීட்சைத் திணைக்களம் இலங்கைப் பரීட்சைத் திணைக்களம் இலங்கைப் பரීட்சைத் திணைக்களம் இலங்கைப் பரීட்சைத் திணைக்களம்  
 Department of Examinations, Sri Lanka Department of Examinations, Sri Lanka Department of Examinations, Sri Lanka Department of Examinations, Sri Lanka Department of Examinations, Sri Lanka  
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 கல்விப் பொதுத் தராதரப் பத்திர (சாதாரண தர)ப் பரீட்சை, 2022(2023)  
 General Certificate of Education (Ord. Level) Examination, 2022(2023)

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 விஞ்ஞானம் I  
 Science I

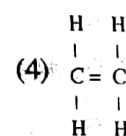
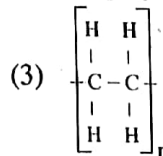
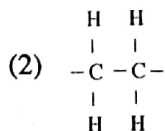
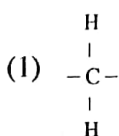
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 ஒரு மணித்தியாலம்.  
 One hour

Note :

- \* Answer all questions.
- \* In each of the questions 1 to 40, pick one of the alternatives (1), (2), (3), (4) which you consider is correct or most appropriate.
- \* Mark a cross (X) on the number corresponding to your choice in the answer sheet provided.
- \* Further instructions are given on the back of the answer sheet. Follow them carefully.

1. What is the organ that mainly performs nitrogenous excretion of human?  
 (1) kidneys (2) lungs (3) skin (4) liver
2. The unit of pressure is,  
 (1)  $N m^{-1}$  (2)  $N m^{-2}$  (3)  $N m$  (4)  $N m^2$
3. What is the type of biomolecules in which hydrogen and oxygen atoms are combined in the ratio of 2 : 1?  
 (1) carbohydrates (2) lipids (3) proteins (4) nucleic acids
4. Select the correct statement given about atoms.  
 (1) The nucleus of all the atoms of every element contains neutrons.  
 (2) In every atom, the number of neutrons and the number of protons in the nucleus are equal.  
 (3) The atomic numbers of two atoms of different elements can be equal.  
 (4) There may be atoms with different mass numbers in the same element.
5. Of the following plant tissues, which is a complex, permanent tissue?  
 (1) parenchyma (2) collenchyma (3) sclerenchyma (4) xylem
6. What is the particle that contributes to the flow of an electric current through a metallic conductor?  
 (1) electron (2) proton (3) neutron (4) metal ion
- Questions 7 and 8 are based on the following chemical equation.  

$$M(s) + ZnSO_4(aq) \longrightarrow MSO_4(aq) + Zn(s)$$
7. In the above chemical equation, the metal indicated as M could be,  
 (1) Cu. (2) Fe. (3) Mg. (4) Pb.
8. What is the type of reaction to which the above chemical equation belongs?  
 (1) combination (2) decomposition (3) single displacement (4) double displacement
9. The kingdoms to which only heterotrophs belong are,  
 (1) fungi and protista. (2) protista and plantae.  
 (3) plantae and animalia. (4) fungi and animalia.
10. Of the following structures, select the repeating unit of polythene.

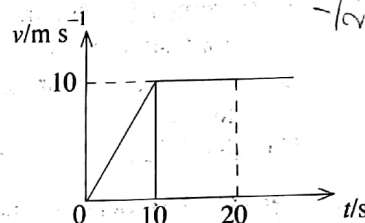


11. Which is the phenomenon that **cannot** be explained by total internal reflection?  
 (1) travelling of light through optical fibres  
 (2) making diamond shiny by cutting and polishing  
 (3) formation of a rainbow by dispersion of white light into colours  
 (4) turning a ray of light by  $90^\circ$  by a rectangular prism
12. Which of the following is a change that occurs after the process of fertilization of a flower?  
 (1) ovary turning into the pericarp (2) ovules turning into seeds  
 (3) sepals turning into the seed coat (4) integument turning into the pericarp
13. Which of the following is **not** a characteristic of the smooth muscle tissue?  
 (1) cells being uninucleate (2) having cross striations  
 (3) cells are spindle shaped (4) acting involuntarily
14. Consider the following statements given about mechanical waves.  
 A - The frequency of the wave changes on entering from one medium into another.  
 B - The speed of the wave does not depend on its frequency.  
 C - The speed of the wave depends on the medium through which it travels.  
 Of the above, true statement/s is/are,  
 (1) only A. (2) only B. (3) only A and C. (4) only B and C.
15. Given below are some information regarding a certain element.  
 • It comes second in the order of abundance of elements in the Earth's crust.  
 • Shows semi-conductor properties.  
 • Shows metallic properties as well as non-metallic properties.  
 This element is,  
 (1) aluminium. (2) silicon. (3) boron. (4) phosphorus.
16. A group of students counted the number of animals in a unit area of an ecosystem. Those numbers are given below.
- | Animal species | butterfly | spider | snail | earthworm | leech | centipede | gecko |
|----------------|-----------|--------|-------|-----------|-------|-----------|-------|
| Number         | 2         | 1      | 3     | 1         | 2     | 1         | 1     |
- What is the number of animals belonging to phylum Annelida in the relevant area?  
 (1) 1 (2) 3 (3) 4 (4) 6
17. Of the following compounds, what is the compound in which the percentage of oxygen by mass is 50%?  
 (H = 1, C = 12, N = 14, O = 16, Mg = 24, Ca = 40)  
 (1)  $\text{NH}_4\text{OH}$  (2)  $\text{Ca}(\text{OH})_2$  (3)  $\text{CH}_3\text{OH}$  (4)  $\text{MgCO}_3$
18. What is the advantage of using the handle fixed to a door in the occasions of opening and closing it?  
 (1) adequacy of applying a low moment of force (2) being able to execute a greater rotation  
 (3) adequacy of applying a smaller force (4) reduction of the amount of work done
19. In the male reproductive system, sperms are produced in the,  
 (1) testes. (2) penis. (3) prostate gland. (4) seminal vesicles.
20. What is the statement that is true with respect to all the molecules given below?  
 $\text{CO}_2$ ,  $\text{NH}_3$ ,  $\text{H}_2\text{O}$   
 (1) The electron octet of the central atom of the molecules is complete.  
 (2) Only single bonds occur among the atoms of the molecules.  
 (3) Lone pairs of electrons are present on the central atom of the molecules.  
 (4) The molecules exist only as gases at room temperature.
21. Consider the following statements given about heat transfer.  
 A - Sea breeze and land breeze occur due to conduction.  
 B - Heat reaches the Earth from the Sun by radiation.  
 C - When a metal spoon is put into a hot cup of tea it is heated by convection.  
 Of the above, true statement/s is/are,  
 (1) only A. (2) only B. (3) only A and C. (4) only B and C.

22. A function of the hormone adrenaline is,  
 (1) preparing body to respond in emergencies.  
 (2) controlling the rate of metabolic activities in the body.  
 (3) stimulating the growth of bones.  
 (4) stimulating the spermatogenesis of males.
23. What is the amount of moles of  $O_2$  in 64 g of oxygen? ( $O = 16$ )  
 (1) 1 (2) 2 (3) 3 (4) 4
24. In a transformer, the voltage supplied to the primary coil is 120 V while the voltage received from the secondary coil is 12 V. If the current flowing through the primary coil is 2 A, what is the current flowing through the secondary coil?  
 (1) 0.2 A (2) 2 A (3) 10 A (4) 20 A
25. Consider the following statements given with regard to anaerobic respiration.  
 A - Lactic acid is produced during the anaerobic respiration taking place in animal cells.  
 B - The amount of energy produced in anaerobic respiration is higher than that in aerobic respiration.  
 C - A part of energy produced during anaerobic respiration is stored as ATP.

Of the above, the true statements are,

- (1) only A and B. (2) only A and C. (3) only B and C. (4) all A, B and C.
26. Indicated here is the velocity-time graph for the motion of a certain object. What is the displacement of the object from the beginning to 20 s?  
 (1) 50 m (2) 100 m  
 (3) 150 m (4) 200 m



27. Consider the following statements.  
 A - In chemical industries, catalysts are used to obtain a higher yield in a shorter period of time.  
 B - Catalysts increase as well as decrease the rate of chemical reactions.

Of the above statements,

- (1) both A and B are true. (2) A is true while B is false.  
 (3) both A and B are false. (4) A is false while B is true.
28. Which of the following phenomena agrees most with Newton's third law of motion?  
 (1) A ball falling down from a higher level bouncing after touching the ground.  
 (2) The velocity of a fruit falling from a tree reaching the maximum value when it comes closer to the ground.  
 (3) Flinging of passengers forward when brakes are suddenly applied to a running bus.  
 (4) Changing the direction of a moving football by kicking it.

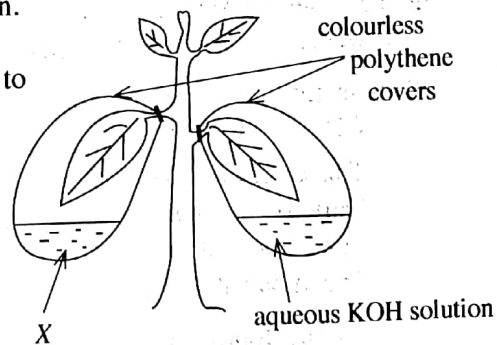
- Questions 29 and 30 below are based on the figure given.

29. The need of which factor for photosynthesis is expected to be tested by this set up?

- (1) water (2) light  
 (3) chlorophyll (4) carbon dioxide

30. What is the substance labelled X in the above figure?

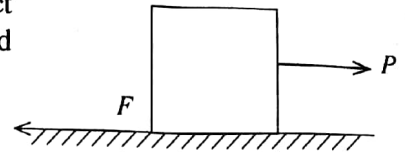
- (1) water (2) lime water  
 (3) iodine solution (4) ethyl alcohol



31. What is the mass of  $C_6H_{12}O_6$  required to prepare  $1 \text{ dm}^3$  of a  $0.1 \text{ mol dm}^{-3}$  glucose solution?

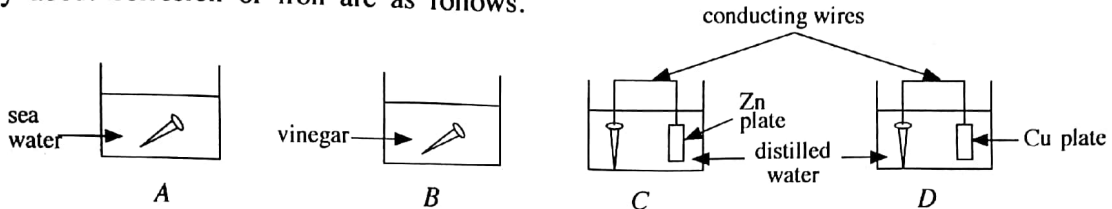
- ( $H = 1, C = 12, O = 16$ )  
 (1) 0.18 g (2) 1.8 g (3) 18 g (4) 180 g

32. The figure shows how a horizontal force  $P$  is applied on an object placed on a plane. When the value of  $P$  is gradually increased from zero, the frictional force ( $F$ ) acting on the object,



- (1) takes a constant value continuously from the beginning.
- (2) increases gradually from zero to a maximum value.
- (3) increases gradually from zero to a maximum value and then decreases gradually.
- (4) increases gradually from zero to a maximum value, then decreases slightly and takes a constant value.

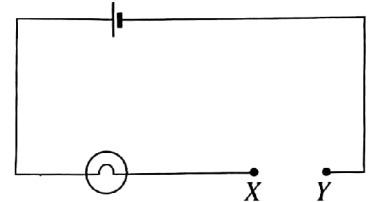
33. Arrangements A, B, C and D set up by a student in the laboratory using four clean, iron nails to study about corrosion of iron are as follows.



When observed after a few days, the nail in which arrangement has corroded least?

- (1) A
- (2) B
- (3) C
- (4) D

34. By connecting a conducting wire between X and Y in the circuit shown in the figure, the bulb in it can be lighted. In three instances, three wires made of the same metal were connected between X and Y as given below and the brightness of the bulb was measured.



- A - connecting a thin long wire
- B - connecting a thick short wire
- C - connecting a thin short wire

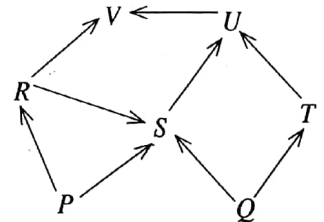
Accordingly, what is the order in which the brightness of the bulb **increases**?

- (1) A , B , C
  - (2) A , C , B
  - (3) B , A , C
  - (4) C , B , A
35. In a water barometer, the vertical height of the water column is 10 m. What is the atmospheric pressure at that instant? (Density of water =  $1000 \text{ kg m}^{-3}$ , Acceleration due to gravity =  $10 \text{ m s}^{-2}$ )
- (1)  $1.0 \times 10^2 \text{ Pa}$
  - (2)  $1.0 \times 10^3 \text{ Pa}$
  - (3)  $1.0 \times 10^4 \text{ Pa}$
  - (4)  $1.0 \times 10^5 \text{ Pa}$

36. What is the chemical compound that ionises completely in aqueous solution and releases  $\text{H}^+$  ions?

- (1)  $\text{CH}_3\text{COOH}$
- (2)  $\text{H}_3\text{PO}_4$
- (3)  $\text{H}_2\text{CO}_3$
- (4)  $\text{HNO}_3$

37. A food web that can be seen in a terrestrial ecosystem is shown in the figure. What is the **most correct** inference that can be made based on this food web?



- (1) R is a herbivore.
- (2) U is a carnivore.
- (3) S is an omnivore.
- (4) V is an omnivore.

38. Of the following, which option indicates the courses of action relevant to sustainable use of resources and use of renewable energy resources respectively?

- (1) waste management and production of electricity by wind power
- (2) reforestation and production of electricity by coal
- (3) converting marshlands into cultivable lands and production of electricity by solar energy
- (4) minimizing food mile and production of electricity by mineral oil

39. Which option indicates the chemical species affecting the environmental crises, ozone layer depletion, acid rains and eutrophication respectively?

- (1) CFC,  $\text{NO}_2$ ,  $\text{CO}_3^{2-}$
- (2) CFC,  $\text{SO}_2$ ,  $\text{NO}_3^-$
- (3)  $\text{NO}_2$ ,  $\text{CO}_2$ ,  $\text{PO}_4^{3-}$
- (4) NO,  $\text{SO}_2$ ,  $\text{SO}_4^{2-}$

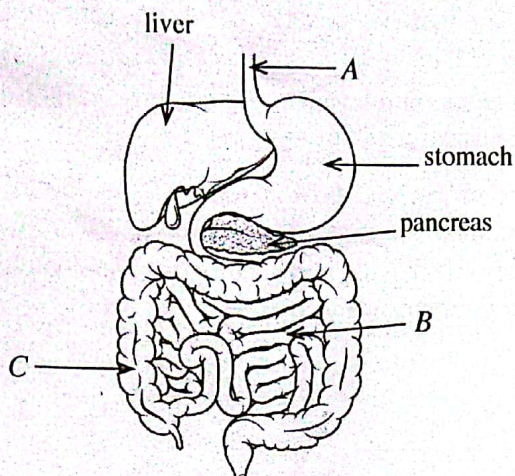
40. Select the pair with inversely proportional relationship.

- (1) concentration of greenhouse gases - atmospheric temperature
- (2) food mile - carbon footprint
- (3) deforestation - desertification
- (4) emergence of invasive species - biodiversity

## Part B

- Answer only **three** questions from the questions No. 5, 6, 7, 8 and 9.

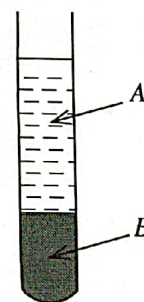
5. (A) Given below is a sketch of the human digestive system.



- Name the parts *A*, *B* and *C*.
- What is the function of the bile produced in the liver in the process of food digestion?
- What is the protein digestive enzyme contained in pancreatic juice?
- State **three** adaptations in structure *B* to absorb digested end products of food to blood efficiently.
- What is the function performed by *C*?
- Inflammation of inner lining of mucosa of stomach is a common disease. What is the name of that disease?

(B) Blood is a specialized connective tissue.

- State **one** function of connective tissues.
- A main feature seen in other connective tissues is not seen in the blood tissue. What is this feature?
- When blood is centrifuged, it separates into two parts as shown in the diagram.
  - In this, by what name is part *A* known?
  - By what name are the nucleated, irregular cells contained in part *B* commonly known?
  - State a function performed by the cells you mentioned in (b) above.



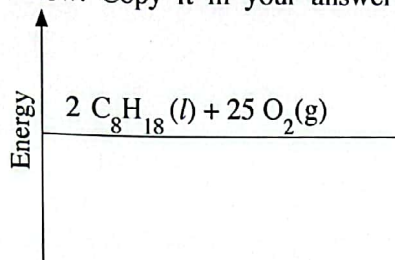
(C) Two systems in the human body operate to maintain coordination and homeostasis. One of them is the nervous system.

- What is the other system relevant to the maintenance of coordination and homeostasis?
- State briefly the meaning of homeostasis.
- What is the structural unit of the nervous system?
- In a reflex action, the pathway taken by the impulses is called the reflex arc. State in the form of a flow chart, the reflex arc from the receptor to the effector in sequential order.

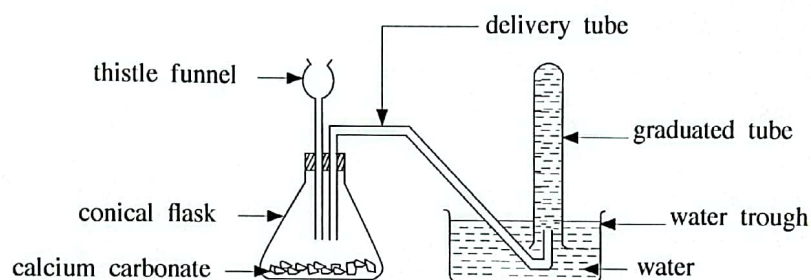
(20 marks)

6. (A) At present light automobiles are run mainly by burning fossil fuels such as petrol (gasoline). The hydrocarbon octane ( $C_8H_{18}$ ) is the main component contained in petrol.
- Explain briefly what hydrocarbons are.
  - (a) Based on the common formula of alkanes, verify that octane is an alkane.  
(b) Name a hydrocarbon that belongs to the alkane series and exists in the gaseous state at room temperature.
  - The balanced chemical equation relevant to the complete combustion of octane is given below.  

$$2 C_8H_{18}(l) + 25 O_2(g) \longrightarrow 16 CO_2(g) + 18 H_2O(l)$$
    - Calculate the mass of carbon dioxide released to the environment by complete combustion of one mole of octane (Relative Molecular Mass of  $CO_2 = 44$ ).
    - An incomplete energy level diagram relating to complete combustion of octane is given below. Copy it in your answer script and complete.



- (B) An apparatus set up by a student to collect a specifically measured volume of carbon dioxide gas using the method of downward displacement of water is given below.

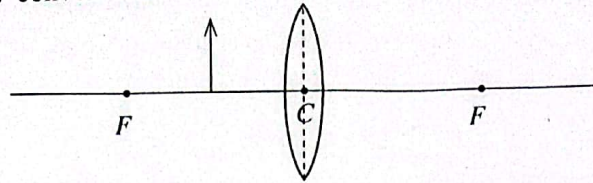


Dilute hydrochloric acid is allowed to drip on the pieces of calcium carbonate through the thistle funnel and carbon dioxide gas is produced by the reaction between those two.

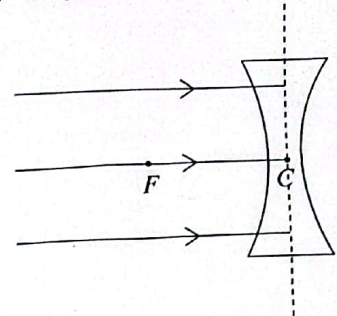
- What change should be made in this set up to prevent the escape of the gas produced through the thistle funnel?
- What is the advantage of using a small conical flask rather than a large conical flask here?
- Write the balanced chemical equation indicating the reaction between calcium carbonate and hydrochloric acid.
- If an equal mass of powdered calcium carbonate was used instead of pieces of calcium carbonate, required volume of the gas can be collected in a shorter period of time. Explain briefly the reason for this.
- (a) Name another method for collecting carbon dioxide gas when the measurement of the volume of the gas is **not** required.  
(b) What physical property of carbon dioxide gas is made use of in the method you stated in part (a) above?
- State a test that can be used in the school laboratory to identify carbon dioxide gas and the observations of it.
- State an instance where non-combustible property of carbon dioxide is made use of.

(20 marks)

7. (A) The following diagram illustrates how an object is placed between the optical centre and the focus of a glass convex lens.



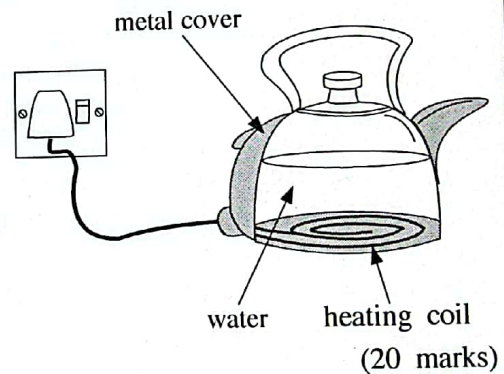
- (i) (a) Copy this diagram in your answer script and draw a ray diagram to construct the image formed.
- (b) State **two** characteristics of that image.
- (ii) (a) Copy the given diagram in your answer script and complete the ray diagram.
- (b) Whatever is the distance of placing an object in front of a concave lens, the image that can be seen has the same characteristics. State **two** characteristics of that image.



- (B) (i) Assume that you are provided with four resistors, each with a resistance of  $2 \Omega$ .
- (a) Draw a diagram indicating, how those can be connected to obtain the highest equivalent resistance.
  - (b) What is the name used to identify the way the resistors are connected as mentioned above?
  - (c) What is the equivalent resistance of that arrangement of resistors?
  - (d) If this arrangement of resistors was connected to a battery with an electromotive force of 8 V, what is the current flowing through the circuit?

(C) The diagram shows a kettle with a heating coil. The kettle contains 1 kg of water.

- (i) What is the main method of transmission of heat produced by the heating coil throughout the entire mass of water?
- (ii) Calculate the amount of heat received by water when water in the kettle was heated from  $25^\circ\text{C}$  to  $50^\circ\text{C}$ . (Specific heat capacity of water =  $4200 \text{ J kg}^{-1} \text{ }^\circ\text{C}^{-1}$ )
- (iii) State **two** properties which a metal used to make the heating coil should necessarily have.
- (iv) When using this kind of a kettle, using a three pin plug is a must. State the reason for it.



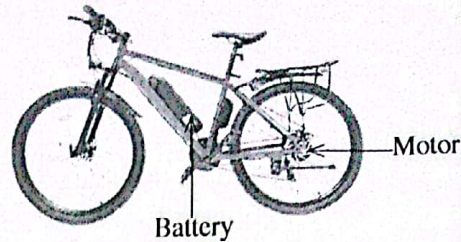
8. (A) A study was conducted by a student on the home garden and the outer environment of his home situated in a rural area. Answer the following questions related to the phenomena identified by him therein.

- (i) Both the papaw plants in the garden had produced flowers. But, of the two plants, always only one plant produced fruits. Explain the reason for this.
- (ii) Though the jasmine plant grown in the garden produces many flowers, those flowers do not produce fruits. Hence, state an artificial vegetative propagation method that could be adopted to obtain a new sapling from a jasmine plant.
- (iii) It was observed that the stamens of the flowers of a *Clerodendrum paniculatum* (pinna/pagoda plant) plant in the garden are turned away from the stigma. What is the importance of that adaptation?
- (iv) When a small animal staying on a plant leaf was observed by a hand lens, jointed appendages and a segmented body could be seen. Name the phylum to which that animal belongs.
- (v) In a fallen plant leaf, the fleshy parts were decayed while the veins were left. A diagram of it is given here.
  - (a) By what name is this venation known?
  - (b) Describe briefly the nature of the root system of the plant with these leaves.



(vi) It was observed that an animal with four limbs staying on a decayed log had a moist skin without scales. What is the class of vertebrates to which this animal belongs?

(B) The diagram illustrates an improved bicycle. It is constructed in such a way that it can be pedalled by a man as well as run by an electric motor.



- (i) (a) Write the transformation of energy that occurs when a man pedals the bicycle.
- (b) Write the transformation of energy that occurs when the bicycle is run by the motor.
- (ii) (a) The voltage supplied to the motor by the battery is 50 V while the maximum power of the motor is 250 W. What is the current drawn from the battery when the motor operates with this power?
- (b) The capacity of the battery is indicated as 10 A h (10 Ampere hours). This means that the battery gets fully discharged in one hour when a current of 10 A is drawn from it. When the current calculated in (a) above is drawn, what time is taken by the battery to discharge completely?
- (c) Find the total distance that can be travelled if the bicycle is to run only by the motor at a constant speed of  $30 \text{ km h}^{-1}$ , at its maximum power, until the fully charged battery becomes completely discharged.
- (iii) Suggest two eco-friendly methods that could be used to charge the battery of this bicycle without using the national electricity grid. (20 marks)

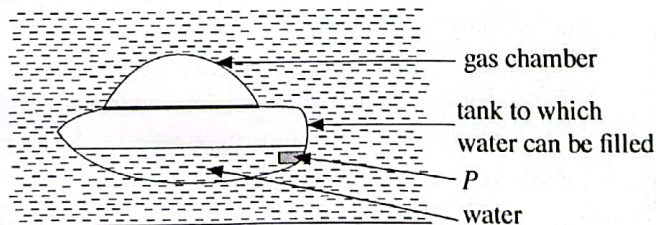
9. (A) Extraction of common salt (sodium chloride) from sea water is a chemical industry carried out in Sri Lanka.

- (i) State **two** environmental factors that should be present in a location suitable to install a saltern.
- (ii) Name the **two** separating techniques relating to the extraction of salt from sea water.
- (iii) Salt taken out from the saltern is heaped as prismatic piles and kept for about six months. What is the reason for this?
- (iv) Some countries in the world obtain drinking water by simple distillation of sea water. Draw a labelled sketch of an apparatus suitable to demonstrate that separating technique in the school laboratory.
- (v) During the electrolysis of acidulated water, it was observed that gas bubbles liberate at the electrodes.
  - (a) At which electrode is a greater volume of gas evolved here?
  - (b) What is the gas evolved at that electrode?

(B) (i) Write Archimedes' principle.

- (ii) When a rubber ball pressed to the bottom of a reservoir with water at rest was released, it could be observed that the ball moved to the surface of water.
  - (a) Action of which force on the ball is the reason for the above observation?
  - (b) State **two** factors on which the magnitude of the force you mentioned above depends.

(iii) The figure illustrates an equipment with a gas chamber and a tank to which water can be filled. Using the device P, water can be filled into and removed from the tank. The figure shows how it stays floating in water when water is filled to a certain volume of the tank.



- (a) Based on the forces acting on the equipment, explain the reason why the equipment stays floating in water.
- (b) What change in the position of the equipment will occur if an extra amount of water is added to the tank?
- (c) In terms of the forces acting on the equipment, explain the reason for the change you stated in (b).
- (d) Name a craft that has been created to travel on the surface of water and inside water using an arrangement such as the one given above. (20 marks)